

## AP Chemistry - HS Science – Year at a Glance

### CScope Order of Units for 2015 - 2016

<b>First Semester</b>		<b>Second Semester</b>	
<b>1<sup>st</sup> Six Weeks</b>	29 days	<b>4<sup>th</sup> Six Weeks</b>	34 days
<b>Chemical Foundations</b> CR5b, CR6, Cr4  <b>Atomic Theory and Atomic Structure</b> BI 1, BI 2, CR5b, CR6  <b>Stoichiometry</b> BI 3, CR5b, CR6		<b>Liquid and Solutions</b> BI 1, BI 2, CR5b, CR6  <b>Solutions</b> BI 2, CR5b, CR6  <b>Kinetics</b> BI 4, CR5b, CR6	
<b>2<sup>nd</sup> Six Weeks</b>	30 days	<b>5<sup>th</sup> Six Weeks</b>	34 days
<b>Reaction Types/ Stoichiometry</b> BI 3, CR5b, CR6  <b>Gases</b> BI 1, BI 2, CR5b, CR6  <b>Thermochemistry</b> BI 5, CR5b, CR6, CR3e		<b>Equilibrium</b> BI 6, CR5b, CR6, CR 3f  <b>Acids and Bases</b> BI 6, CR5b, CR6  <b>Aqueous Equilibrium</b> BI 6, CR5b, CR6	
<b>3<sup>rd</sup> Six Weeks</b>	23 days	<b>6<sup>th</sup> Six Weeks</b>	28 days
<b>Atomic Structure / Periodicity</b> BI 1, BI 2, CR5b, CR6, CR 3a  <b>Chemical Bonding</b> BI 1, BI 2, CR5b, CR6, CR3b  <b>Covalent Bonding</b> BI 1, BI 2, CR5b, CR6		<b>Thermodynamics</b> BI 5, CR5b, CR6  <b>Electrochemistry</b> BI 3, CR5b, CR6  <b>Nuclear / Representative Elements</b> BI 2, CR5b, CR6	

Updated: 8/25/15